

ADDITIONS AND CORRECTIONS

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Murthy S. Gudipati*[†] and **Daniel H. Katayama***[‡]: Corrected Rate Constants for Collision-Induced Electronic Transitions from the N₂ a¹Π_g (*v* = 0 and 1) Levels.

Pages 7854–7858. Several years ago, one of the present authors (DHK) and colleagues published two papers^{1,2} on the collision-induced electronic transitions (CIET) between the nested vibronic levels of the electronic states a¹Π_g and a'¹Σ_u⁻ of N₂. Using a double-resonance-enhanced multiphoton ionization (REMPI) technique, rate constants and branching ratios between the adjacent vibrational levels of the *a* and *a'* states were determined in these studies.

Recently, one of us (MSG) found an error in the straightforward numerical analysis of eq 8 in ref 2. This arithmetical error led to incorrect rate constants being reported for N₂a(*v* = 1) in Table 1 of this reference. We correct these errors in the present work because of the important role played by these rate constants in modeling the earth's upper atmosphere.³

With the correct values for eq 8, eq 14 in ref 2 will become

$$A(t) = A'_0 \{ 19.23 \exp(-83.8k_{1 \rightarrow 1}[Q]t) + 63.8 \exp(-0.77k_{1 \rightarrow 1}[Q]t) \} \quad (1)$$

and the second term dominates as a single exponential function, resulting in eq 16 of this reference being written as

$$1/\tau = 0.77k_{1 \rightarrow 1}[Q] \quad (2)$$

Using these equations, the rate constants and cross sections for N₂a(*v* = 1) given in Table 1 of ref 2 are corrected and

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TABLE 1: State-Specific Rate Constants and Cross Sections for Quenching the N₂ a¹Π_g(*v* = 0 and 1) Levels as a Function of Collision Partner at 295 K^a

data for <i>v</i> = 1						
collision partner	<i>k</i> _{1→2} ^b	σ _{1→2}	<i>k</i> _{2→1} ^b	σ _{2→1}	<i>k</i> _{1→1} ^c	σ _{1→1}
helium	7.8	5.8	25.3	19.5	0.39	0.29
argon	19.5	33	62	103	0.97	1.56
nitrogen	48.7	74	164	245	2.53	3.9
data for <i>v</i> = 0						
collision partner	<i>k</i> _{0→1} ^c	σ _{0→1}	<i>k</i> _{1→0} ^c	σ _{1→0}	<i>k</i> _{0→0} ^c	σ _{0→0}
helium	0.5	0.4	3.9	3.0	0.7	0.5
argon	0.8	1.3	6.6	10.8	1.2	2.0
nitrogen	1.7	2.6	14.3	21.5	2.6	3.9

^a The rate constants, *k*, are in units of 10⁻¹¹ molecule⁻¹ s⁻¹ cm³, and the cross sections, σ, are in 10⁻¹⁶ cm². ^b The estimated error for these rate constants is ±1. ^c The estimated error for this rate constant is ±0.1.

summarized in Table 1 here. We also include the rate constants and cross sections for N₂a(*v* = 0) at room temperature given in Table 1 of ref 1, for which erroneous values of *k*_{1→0} and σ_{1→0} have been corrected.

Finally, in section IV (Discussion and Conclusions) of ref 2, there are typographical errors. The second sentence should read "... deactivation path from a(*v* = 1) is to the a'(*v* = 2 and 1) levels, respectively." Similarly, the fourth sentence should read, "The branching ratio from a(*v* = 1) to a'(*v* = 2 and 1) is independent of these collision partners..."

References and Notes

- (1) Katayama, D. H.; Dentamaro, A. V.; Welsh, J. A. *J. Chem. Phys.* **1984**, *101*, 9422–9428.
- (2) Katayama, D. H.; Dentamaro, A. V.; Welsh, J. A. *J. Phys. Chem.* **1996**, *100*, 7854–7858.
- (3) Eastes, R. W. *J. Geophys. Res.* **2000**, *105*, 18557–18573.

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